

### Atomic Masses (amu)

| <i>IA</i>                      | <i>IIA</i>                | <i>IIIB</i>               | <i>IVB</i>                 | <i>VB</i>                  | <i>VIIB</i>                | <i>VIIIB</i>               | <i>VIIIB</i>               | <i>IB</i>                  | <i>IIB</i>                | <i>IIIA</i>               | <i>IVA</i>                | <i>VA</i>                 | <i>VIA</i>                | <i>VIIA</i>               | <i>VIIIA</i>              |                           |                           |
|--------------------------------|---------------------------|---------------------------|----------------------------|----------------------------|----------------------------|----------------------------|----------------------------|----------------------------|---------------------------|---------------------------|---------------------------|---------------------------|---------------------------|---------------------------|---------------------------|---------------------------|---------------------------|
| 1                              | 2                         | 3                         | 4                          | 5                          | 6                          | 7                          | 8                          | 9                          | 10                        | 11                        | 12                        | 13                        | 14                        | 15                        | 16                        | 17                        | 18                        |
| 1<br><b>H</b><br>1.008         |                           |                           |                            |                            |                            |                            |                            |                            |                           |                           |                           |                           |                           |                           |                           |                           | 2<br><b>He</b><br>4.003   |
| 2<br>3<br><b>Li</b><br>6.941   | 4<br><b>Be</b><br>9.012   |                           |                            |                            |                            |                            |                            |                            |                           |                           |                           |                           |                           |                           |                           |                           | 10<br><b>Ne</b><br>20.18  |
| 11<br>3<br><b>Na</b><br>22.99  | 12<br><b>Mg</b><br>24.31  |                           |                            |                            |                            |                            |                            |                            |                           |                           |                           |                           |                           |                           |                           |                           | 18<br><b>Ar</b><br>39.95  |
| 19<br>4<br><b>K</b><br>39.10   | 20<br><b>Ca</b><br>40.08  | 21<br><b>Sc</b><br>44.96  | 22<br><b>Ti</b><br>47.87   | 23<br><b>V</b><br>50.94    | 24<br><b>Cr</b><br>52.00   | 25<br><b>Mn</b><br>54.94   | 26<br><b>Fe</b><br>55.85   | 27<br><b>Co</b><br>58.93   | 28<br><b>Ni</b><br>58.69  | 29<br><b>Cu</b><br>63.55  | 30<br><b>Zn</b><br>65.41  | 31<br><b>Ga</b><br>69.72  | 32<br><b>Ge</b><br>72.64  | 33<br><b>As</b><br>74.92  | 34<br><b>Se</b><br>78.96  | 35<br><b>Br</b><br>79.90  | 36<br><b>Kr</b><br>83.80  |
| 37<br>5<br><b>Rb</b><br>85.47  | 38<br><b>Sr</b><br>87.62  | 39<br><b>Y</b><br>88.91   | 40<br><b>Zr</b><br>91.22   | 41<br><b>Nb</b><br>92.91   | 42<br><b>Mo</b><br>95.94   | 43<br><b>Tc</b><br>97.91   | 44<br><b>Ru</b><br>101.07  | 45<br><b>Rh</b><br>102.91  | 46<br><b>Pd</b><br>106.42 | 47<br><b>Ag</b><br>107.87 | 48<br><b>Cd</b><br>112.41 | 49<br><b>In</b><br>114.82 | 50<br><b>Sn</b><br>118.71 | 51<br><b>Sb</b><br>121.76 | 52<br><b>Te</b><br>127.60 | 53<br><b>I</b><br>126.90  | 54<br><b>Xe</b><br>131.29 |
| 55<br>6<br><b>Cs</b><br>132.91 | 56<br><b>Ba</b><br>137.33 | 57<br><b>La</b><br>138.91 | 72<br><b>Hf</b><br>178.49  | 73<br><b>Ta</b><br>180.95  | 74<br><b>W</b><br>183.84   | 75<br><b>Re</b><br>186.21  | 76<br><b>Os</b><br>190.23  | 77<br><b>Ir</b><br>192.22  | 78<br><b>Pt</b><br>195.08 | 79<br><b>Au</b><br>196.97 | 80<br><b>Hg</b><br>200.59 | 81<br><b>Tl</b><br>204.38 | 82<br><b>Pb</b><br>207.21 | 83<br><b>Bi</b><br>208.98 | 84<br><b>Po</b><br>208.98 | 85<br><b>At</b><br>209.99 | 86<br><b>Rn</b><br>222.02 |
| 87<br>7<br><b>Fr</b><br>223.02 | 88<br><b>Ra</b><br>226.03 | 89<br><b>Ac</b><br>227.03 | 104<br><b>Rf</b><br>261.11 | 105<br><b>Db</b><br>262.11 | 106<br><b>Sg</b><br>266.12 | 107<br><b>Bh</b><br>264.12 | 108<br><b>Hs</b><br>269.13 | 109<br><b>Mt</b><br>268.24 | 110<br><b>Ds</b><br>271   | 111<br><b>Rg</b><br>272   | 112<br><b>Cn</b><br>275   |                           |                           |                           |                           |                           |                           |

|   |                           |                           |                           |                           |                           |                           |                           |                           |                           |                           |                            |                            |                            |                            |
|---|---------------------------|---------------------------|---------------------------|---------------------------|---------------------------|---------------------------|---------------------------|---------------------------|---------------------------|---------------------------|----------------------------|----------------------------|----------------------------|----------------------------|
| 6 | 58<br><b>Ce</b><br>140.12 | 59<br><b>Pr</b><br>140.91 | 60<br><b>Nd</b><br>144.24 | 61<br><b>Pm</b><br>144.91 | 62<br><b>Sm</b><br>150.36 | 63<br><b>Eu</b><br>151.96 | 64<br><b>Gd</b><br>157.25 | 65<br><b>Tb</b><br>158.93 | 66<br><b>Dy</b><br>162.50 | 67<br><b>Ho</b><br>164.93 | 68<br><b>Er</b><br>167.26  | 69<br><b>Tm</b><br>168.93  | 70<br><b>Yb</b><br>173.04  | 71<br><b>Lu</b><br>174.97  |
|   | 90<br><b>Th</b><br>232.04 | 91<br><b>Pa</b><br>231.04 | 92<br><b>U</b><br>238.03  | 93<br><b>Np</b><br>237.05 | 94<br><b>Pu</b><br>244.06 | 95<br><b>Am</b><br>243.06 | 96<br><b>Cm</b><br>247.07 | 97<br><b>Bk</b><br>247.07 | 98<br><b>Cf</b><br>251.08 | 99<br><b>Es</b><br>252.08 | 100<br><b>Fm</b><br>257.10 | 101<br><b>Md</b><br>258.10 | 102<br><b>No</b><br>259.10 | 103<br><b>Lr</b><br>262.11 |

**Element Names**

| <i>IA</i>                  | <i>IIA</i>                   | <i>IIIB</i>                  | <i>IVB</i>                        | <i>VB</i>                   | <i>VIB</i>                     | <i>VIIB</i>                   | <i>VIIIB</i>                 | <i>VIIIB</i>                   | <i>IB</i>                        | <i>IIB</i>                      | <i>IIIA</i>                     | <i>IVA</i>                  | <i>VA</i>                    | <i>VIA</i>                  | <i>VIIA</i>                  | <i>VIIIA</i>                |                            |                              |                          |                             |                          |
|----------------------------|------------------------------|------------------------------|-----------------------------------|-----------------------------|--------------------------------|-------------------------------|------------------------------|--------------------------------|----------------------------------|---------------------------------|---------------------------------|-----------------------------|------------------------------|-----------------------------|------------------------------|-----------------------------|----------------------------|------------------------------|--------------------------|-----------------------------|--------------------------|
| 1                          | 2                            | 3                            | 4                                 | 5                           | 6                              | 7                             | 8                            | 9                              | 10                               | 11                              | 12                              | 13                          | 14                           | 15                          | 16                           | 17                          | 18                         |                              |                          |                             |                          |
| 1<br><b>H</b><br>hydrogen  |                              |                              |                                   |                             |                                |                               |                              |                                |                                  |                                 |                                 |                             |                              |                             |                              | 2<br><b>He</b><br>helium    |                            |                              |                          |                             |                          |
| 2<br><b>Li</b><br>lithium  | 4<br><b>Be</b><br>beryllium  |                              |                                   |                             |                                |                               |                              |                                |                                  |                                 |                                 |                             |                              |                             |                              | 5<br><b>B</b><br>boron      | 6<br><b>C</b><br>carbon    | 7<br><b>N</b><br>nitrogen    | 8<br><b>O</b><br>oxygen  | 9<br><b>F</b><br>fluorine   | 10<br><b>Ne</b><br>neon  |
| 3<br><b>Na</b><br>sodium   | 11<br><b>Mg</b><br>magnesium |                              |                                   |                             |                                |                               |                              |                                |                                  |                                 |                                 |                             |                              |                             |                              | 13<br><b>Al</b><br>aluminum | 14<br><b>Si</b><br>silicon | 15<br><b>P</b><br>phosphorus | 16<br><b>S</b><br>sulfur | 17<br><b>Cl</b><br>chlorine | 18<br><b>Ar</b><br>argon |
| 4<br><b>K</b><br>potassium | 20<br><b>Ca</b><br>calcium   | 21<br><b>Sc</b><br>scandium  | 22<br><b>Ti</b><br>titanium       | 23<br><b>V</b><br>vanadium  | 24<br><b>Cr</b><br>chromium    | 25<br><b>Mn</b><br>manganese  | 26<br><b>Fe</b><br>iron      | 27<br><b>Co</b><br>cobalt      | 28<br><b>Ni</b><br>nickel        | 29<br><b>Cu</b><br>copper       | 30<br><b>Zn</b><br>zinc         | 31<br><b>Ga</b><br>gallium  | 32<br><b>Ge</b><br>germanium | 33<br><b>As</b><br>arsenic  | 34<br><b>Se</b><br>selenium  | 35<br><b>Br</b><br>bromine  | 36<br><b>Kr</b><br>krypton |                              |                          |                             |                          |
| 5<br><b>Rb</b><br>rubidium | 38<br><b>Sr</b><br>strontium | 39<br><b>Y</b><br>yttrium    | 40<br><b>Zr</b><br>zirconium      | 41<br><b>Nb</b><br>niobium  | 42<br><b>Mo</b><br>molybdenum  | 43<br><b>Tc</b><br>technetium | 44<br><b>Ru</b><br>ruthenium | 45<br><b>Rh</b><br>rhodium     | 46<br><b>Pd</b><br>palladium     | 47<br><b>Ag</b><br>silver       | 48<br><b>Cd</b><br>cadmium      | 49<br><b>In</b><br>indium   | 50<br><b>Sn</b><br>tin       | 51<br><b>Sb</b><br>antimony | 52<br><b>Te</b><br>tellurium | 53<br><b>I</b><br>iodine    | 54<br><b>Xe</b><br>xenon   |                              |                          |                             |                          |
| 6<br><b>Cs</b><br>cesium   | 56<br><b>Ba</b><br>barium    | 57<br><b>La</b><br>lanthanum | 72<br><b>Hf</b><br>hafnium        | 73<br><b>Ta</b><br>tantalum | 74<br><b>W</b><br>tungsten     | 75<br><b>Re</b><br>rhenum     | 76<br><b>Os</b><br>osmium    | 77<br><b>Ir</b><br>iridium     | 78<br><b>Pt</b><br>platinum      | 79<br><b>Au</b><br>gold         | 80<br><b>Hg</b><br>mercury      | 81<br><b>Tl</b><br>thallium | 82<br><b>Pb</b><br>lead      | 83<br><b>Bi</b><br>bismuth  | 84<br><b>Po</b><br>polonium  | 85<br><b>At</b><br>astatine | 86<br><b>Rn</b><br>radon   |                              |                          |                             |                          |
| 7<br><b>Fr</b><br>francium | 88<br><b>Ra</b><br>radium    | 89<br><b>Ac</b><br>actinium  | 104<br><b>Rf</b><br>rutherfordium | 105<br><b>Db</b><br>dubnium | 106<br><b>Sg</b><br>seaborgium | 107<br><b>Bh</b><br>bohrium   | 108<br><b>Hs</b><br>hassium  | 109<br><b>Mt</b><br>meitnerium | 110<br><b>Ds</b><br>darmstadtium | 111<br><b>Rg</b><br>roentgenium | 112<br><b>Cn</b><br>copernicium |                             |                              |                             |                              |                             |                            |                              |                          |                             |                          |

|                           |                                 |                              |                               |                              |                              |                               |                              |                                |                                |                            |                                 |                              |                                |
|---------------------------|---------------------------------|------------------------------|-------------------------------|------------------------------|------------------------------|-------------------------------|------------------------------|--------------------------------|--------------------------------|----------------------------|---------------------------------|------------------------------|--------------------------------|
| 6<br><b>Ce</b><br>cerium  | 58<br><b>Pr</b><br>praseodymium | 59<br><b>Nd</b><br>neodymium | 60<br><b>Pm</b><br>promethium | 61<br><b>Sm</b><br>samarium  | 62<br><b>Eu</b><br>europium  | 63<br><b>Gd</b><br>gadolinium | 64<br><b>Tb</b><br>terbium   | 65<br><b>Dy</b><br>dysprosium  | 66<br><b>Ho</b><br>holmium     | 67<br><b>Er</b><br>erbium  | 68<br><b>Tm</b><br>thulium      | 69<br><b>Yb</b><br>ytterbium | 70<br><b>Lu</b><br>lutetium    |
| 7<br><b>Th</b><br>thorium | 90<br><b>Pa</b><br>protactinium | 91<br><b>U</b><br>uranium    | 92<br><b>Np</b><br>neptunium  | 93<br><b>Pu</b><br>plutonium | 94<br><b>Am</b><br>americium | 95<br><b>Cm</b><br>curium     | 96<br><b>Bk</b><br>berkelium | 97<br><b>Cf</b><br>californium | 98<br><b>Es</b><br>einsteinium | 99<br><b>Fm</b><br>fermium | 100<br><b>Md</b><br>mendelevium | 101<br><b>No</b><br>nobelium | 102<br><b>Lr</b><br>lawrencium |

### Classification of the Elements

| <i>IA</i>                   | <i>IIA</i>                  | <i>IIIB</i>                 | <i>IVB</i>                   | <i>VB</i>                    | <i>VIB</i>                   | <i>VIIB</i>                  | <i>VIIIB</i>                 | <i>VIIIB</i>                 | <i>IB</i>                    | <i>IIB</i>                   | <i>IIIA</i>                  | <i>IVA</i>                  | <i>VA</i>                    | <i>VIA</i>                   | <i>VIIA</i>                  | <i>VIIIA</i>                 |                              |
|-----------------------------|-----------------------------|-----------------------------|------------------------------|------------------------------|------------------------------|------------------------------|------------------------------|------------------------------|------------------------------|------------------------------|------------------------------|-----------------------------|------------------------------|------------------------------|------------------------------|------------------------------|------------------------------|
| 1                           | 2                           | 3                           | 4                            | 5                            | 6                            | 7                            | 8                            | 9                            | 10                           | 11                           | 12                           | 13                          | 14                           | 15                           | 16                           | 17                           | 18                           |
| 1<br><b>H</b><br>non-metal  |                             |                             |                              |                              |                              |                              |                              |                              |                              |                              |                              |                             |                              |                              |                              |                              | 2<br><b>He</b><br>non-metal  |
| 3<br><b>Li</b><br>metallic  | 4<br><b>Be</b><br>metallic  |                             |                              |                              |                              |                              |                              |                              |                              |                              |                              |                             |                              |                              |                              |                              | 5<br><b>B</b><br>metalloid   |
| 11<br><b>Na</b><br>metallic | 12<br><b>Mg</b><br>metallic |                             |                              |                              |                              |                              |                              |                              |                              |                              |                              |                             |                              |                              |                              |                              | 6<br><b>C</b><br>non-metal   |
| 19<br><b>K</b><br>metallic  | 20<br><b>Ca</b><br>metallic | 21<br><b>Sc</b><br>metallic | 22<br><b>Ti</b><br>metallic  | 23<br><b>V</b><br>metallic   | 24<br><b>Cr</b><br>metallic  | 25<br><b>Mn</b><br>metallic  | 26<br><b>Fe</b><br>metallic  | 27<br><b>Co</b><br>metallic  | 28<br><b>Ni</b><br>metallic  | 29<br><b>Cu</b><br>metallic  | 30<br><b>Zn</b><br>metallic  | 31<br><b>Ga</b><br>metallic | 32<br><b>Ge</b><br>metalloid | 33<br><b>As</b><br>non-metal | 34<br><b>Se</b><br>non-metal | 35<br><b>Br</b><br>non-metal | 36<br><b>Kr</b><br>non-metal |
| 37<br><b>Rb</b><br>metallic | 38<br><b>Sr</b><br>metallic | 39<br><b>Y</b><br>metallic  | 40<br><b>Zr</b><br>metallic  | 41<br><b>Nb</b><br>metallic  | 42<br><b>Mo</b><br>metallic  | 43<br><b>Tc</b><br>metallic  | 44<br><b>Ru</b><br>metallic  | 45<br><b>Rh</b><br>metallic  | 46<br><b>Pd</b><br>metallic  | 47<br><b>Ag</b><br>metallic  | 48<br><b>Cd</b><br>metallic  | 49<br><b>In</b><br>metallic | 50<br><b>Sn</b><br>metalloid | 51<br><b>Sb</b><br>metalloid | 52<br><b>Te</b><br>non-metal | 53<br><b>I</b><br>non-metal  | 54<br><b>Xe</b><br>non-metal |
| 55<br><b>Cs</b><br>metallic | 56<br><b>Ba</b><br>metallic | 57<br><b>La</b><br>metallic | 72<br><b>Hf</b><br>metallic  | 73<br><b>Ta</b><br>metallic  | 74<br><b>W</b><br>metallic   | 75<br><b>Re</b><br>metallic  | 76<br><b>Os</b><br>metallic  | 77<br><b>Ir</b><br>metallic  | 78<br><b>Pt</b><br>metallic  | 79<br><b>Au</b><br>metallic  | 80<br><b>Hg</b><br>metallic  | 81<br><b>Tl</b><br>metallic | 82<br><b>Pb</b><br>metallic  | 83<br><b>Bi</b><br>metallic  | 84<br><b>Po</b><br>metallic  | 85<br><b>At</b><br>non-metal | 86<br><b>Rn</b><br>non-metal |
| 87<br><b>Fr</b><br>metallic | 88<br><b>Ra</b><br>metallic | 89<br><b>Ac</b><br>metallic | 104<br><b>Rf</b><br>metallic | 105<br><b>Db</b><br>metallic | 106<br><b>Sg</b><br>metallic | 107<br><b>Bh</b><br>metallic | 108<br><b>Hs</b><br>metallic | 109<br><b>Mt</b><br>metallic | 110<br><b>Ds</b><br>metallic | 111<br><b>Rg</b><br>metallic | 112<br><b>Cn</b><br>metallic |                             |                              |                              |                              |                              |                              |

|   |                             |                             |                             |                             |                             |                             |                             |                             |                             |                             |                              |                              |                              |                              |
|---|-----------------------------|-----------------------------|-----------------------------|-----------------------------|-----------------------------|-----------------------------|-----------------------------|-----------------------------|-----------------------------|-----------------------------|------------------------------|------------------------------|------------------------------|------------------------------|
| 6 | 58<br><b>Ce</b><br>metallic | 59<br><b>Pr</b><br>metallic | 60<br><b>Nd</b><br>metallic | 61<br><b>Pm</b><br>metallic | 62<br><b>Sm</b><br>metallic | 63<br><b>Eu</b><br>metallic | 64<br><b>Gd</b><br>metallic | 65<br><b>Tb</b><br>metallic | 66<br><b>Dy</b><br>metallic | 67<br><b>Ho</b><br>metallic | 68<br><b>Er</b><br>metallic  | 69<br><b>Tm</b><br>metallic  | 70<br><b>Yb</b><br>metallic  | 71<br><b>Lu</b><br>metallic  |
|   | 90<br><b>Th</b><br>metallic | 91<br><b>Pa</b><br>metallic | 92<br><b>U</b><br>metallic  | 93<br><b>Np</b><br>metallic | 94<br><b>Pu</b><br>metallic | 95<br><b>Am</b><br>metallic | 96<br><b>Cm</b><br>metallic | 97<br><b>Bk</b><br>metallic | 98<br><b>Cf</b><br>metallic | 99<br><b>Es</b><br>metallic | 100<br><b>Fm</b><br>metallic | 101<br><b>Md</b><br>metallic | 102<br><b>No</b><br>metallic | 103<br><b>Lr</b><br>metallic |

### Group (Family) Names

| <i>IA</i>       | <i>IIA</i>      | <i>IIIB</i>     | <i>IVB</i>       | <i>VB</i>        | <i>VIIB</i>      | <i>VIIB</i>      | <i>VIIIB</i>     | <i>VIIIB</i>     | <i>IB</i>        | <i>IIB</i>       | <i>IIIA</i>      | <i>IVA</i>      | <i>VA</i>       | <i>VIA</i>      | <i>VIIA</i>     | <i>VIIIA</i>    |                 |
|-----------------|-----------------|-----------------|------------------|------------------|------------------|------------------|------------------|------------------|------------------|------------------|------------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|
| 1               | 2               | 3               | 4                | 5                | 6                | 7                | 8                | 9                | 10               | 11               | 12               | 13              | 14              | 15              | 16              | 17              | 18              |
| 1<br><b>H</b>   |                 |                 |                  |                  |                  |                  |                  |                  |                  |                  |                  |                 |                 |                 |                 |                 | 2<br><b>He</b>  |
| 3<br><b>Li</b>  | 4<br><b>Be</b>  |                 |                  |                  |                  |                  |                  |                  |                  |                  |                  |                 |                 |                 |                 |                 | 5<br><b>B</b>   |
| 11<br><b>Na</b> | 12<br><b>Mg</b> |                 |                  |                  |                  |                  |                  |                  |                  |                  |                  |                 |                 |                 |                 |                 | 6<br><b>C</b>   |
| 19<br><b>K</b>  | 20<br><b>Ca</b> | 21<br><b>Sc</b> | 22<br><b>Ti</b>  | 23<br><b>V</b>   | 24<br><b>Cr</b>  | 25<br><b>Mn</b>  | 26<br><b>Fe</b>  | 27<br><b>Co</b>  | 28<br><b>Ni</b>  | 29<br><b>Cu</b>  | 30<br><b>Zn</b>  | 31<br><b>Ga</b> | 32<br><b>Ge</b> | 33<br><b>As</b> | 34<br><b>Se</b> | 35<br><b>Br</b> | 36<br><b>Kr</b> |
| 37<br><b>Rb</b> | 38<br><b>Sr</b> | 39<br><b>Y</b>  | 40<br><b>Zr</b>  | 41<br><b>Nb</b>  | 42<br><b>Mo</b>  | 43<br><b>Tc</b>  | 44<br><b>Ru</b>  | 45<br><b>Rh</b>  | 46<br><b>Pd</b>  | 47<br><b>Ag</b>  | 48<br><b>Cd</b>  | 49<br><b>In</b> | 50<br><b>Sn</b> | 51<br><b>Sb</b> | 52<br><b>Te</b> | 53<br><b>I</b>  | 54<br><b>Xe</b> |
| 55<br><b>Cs</b> | 56<br><b>Ba</b> | 57<br><b>La</b> | 72<br><b>Hf</b>  | 73<br><b>Ta</b>  | 74<br><b>W</b>   | 75<br><b>Re</b>  | 76<br><b>Os</b>  | 77<br><b>Ir</b>  | 78<br><b>Pt</b>  | 79<br><b>Au</b>  | 80<br><b>Hg</b>  | 81<br><b>Tl</b> | 82<br><b>Pb</b> | 83<br><b>Bi</b> | 84<br><b>Po</b> | 85<br><b>At</b> | 86<br><b>Rn</b> |
| 87<br><b>Fr</b> | 88<br><b>Ra</b> | 89<br><b>Ac</b> | 104<br><b>Rf</b> | 105<br><b>Db</b> | 106<br><b>Sg</b> | 107<br><b>Bh</b> | 108<br><b>Hs</b> | 109<br><b>Mt</b> | 110<br><b>Ds</b> | 111<br><b>Rg</b> | 112<br><b>Cn</b> |                 |                 |                 |                 |                 |                 |

group 1: alkali metals (except H)

group 2: alkaline earth metals

groups 3-12: outer transition metals

group 13: boron family

group 14: carbon family

group 15: nitrogen (phosphorus) family

group 16: oxygen (chalcogen) family

group 17: halogen family

group 18: noble gas family

lanthanoid and actinoid series: inner transition metals

|                 |                 |                 |                 |                 |                 |                 |                 |                 |                 |                  |                  |                  |                  |
|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|------------------|------------------|------------------|------------------|
| 58<br><b>Ce</b> | 59<br><b>Pr</b> | 60<br><b>Nd</b> | 61<br><b>Pm</b> | 62<br><b>Sm</b> | 63<br><b>Eu</b> | 64<br><b>Gd</b> | 65<br><b>Tb</b> | 66<br><b>Dy</b> | 67<br><b>Ho</b> | 68<br><b>Er</b>  | 69<br><b>Tm</b>  | 70<br><b>Yb</b>  | 71<br><b>Lu</b>  |
| 90<br><b>Th</b> | 91<br><b>Pa</b> | 92<br><b>U</b>  | 93<br><b>Np</b> | 94<br><b>Pu</b> | 95<br><b>Am</b> | 96<br><b>Cm</b> | 97<br><b>Bk</b> | 98<br><b>Cf</b> | 99<br><b>Es</b> | 100<br><b>Fm</b> | 101<br><b>Md</b> | 102<br><b>No</b> | 103<br><b>Lr</b> |

## Electronegativity (Pauling)

| <i>IA</i>              | <i>IIA</i>              | <i>IIIB</i>             | <i>IVB</i>               | <i>VB</i>                | <i>VIIB</i>              | <i>VIIIB</i>             | <i>VIIIB</i>             | <i>VIIIB</i>             | <i>IB</i>                | <i>IIB</i>               | <i>IIIA</i>              | <i>IVA</i>              | <i>VA</i>               | <i>VIA</i>              | <i>VIIA</i>             | <i>VIIIA</i>            |                         |
|------------------------|-------------------------|-------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|-------------------------|-------------------------|-------------------------|-------------------------|-------------------------|-------------------------|
| 1                      | 2                       | 3                       | 4                        | 5                        | 6                        | 7                        | 8                        | 9                        | 10                       | 11                       | 12                       | 13                      | 14                      | 15                      | 16                      | 17                      | 18                      |
| 1<br><b>H</b><br>2.20  |                         |                         |                          |                          |                          |                          |                          |                          |                          |                          |                          |                         |                         |                         |                         |                         | 2<br><b>He</b>          |
| 2<br><b>Li</b><br>0.98 | 3<br><b>Be</b><br>1.57  |                         |                          |                          |                          |                          |                          |                          |                          |                          |                          |                         |                         |                         |                         |                         | 10<br><b>Ne</b>         |
| 2<br><b>Na</b><br>0.93 | 11<br><b>Mg</b><br>1.31 |                         |                          |                          |                          |                          |                          |                          |                          |                          |                          |                         |                         |                         |                         |                         | 18<br><b>Ar</b>         |
| 4<br><b>K</b><br>0.82  | 20<br><b>Ca</b><br>1.00 | 21<br><b>Sc</b><br>1.36 | 22<br><b>Ti</b><br>1.54  | 23<br><b>V</b><br>1.63   | 24<br><b>Cr</b><br>1.66  | 25<br><b>Mn</b><br>1.55  | 26<br><b>Fe</b><br>1.83  | 27<br><b>Co</b><br>1.88  | 28<br><b>Ni</b><br>1.91  | 29<br><b>Cu</b><br>1.90  | 30<br><b>Zn</b><br>1.65  | 31<br><b>Ga</b><br>1.81 | 32<br><b>Ge</b><br>2.01 | 33<br><b>As</b><br>2.18 | 34<br><b>Se</b><br>2.55 | 35<br><b>Br</b><br>2.96 | 36<br><b>Kr</b><br>3.00 |
| 5<br><b>Rb</b><br>0.82 | 38<br><b>Sr</b><br>0.95 | 39<br><b>Y</b><br>1.22  | 40<br><b>Zr</b><br>1.33  | 41<br><b>Nb</b><br>1.60  | 42<br><b>Mo</b><br>2.16  | 43<br><b>Tc</b><br>2.10  | 44<br><b>Ru</b><br>2.20  | 45<br><b>Rh</b><br>2.28  | 46<br><b>Pd</b><br>2.20  | 47<br><b>Ag</b><br>1.93  | 48<br><b>Cd</b><br>1.69  | 49<br><b>In</b><br>1.78 | 50<br><b>Sn</b><br>1.96 | 51<br><b>Sb</b><br>2.05 | 52<br><b>Te</b><br>2.10 | 53<br><b>I</b><br>2.66  | 54<br><b>Xe</b><br>2.60 |
| 6<br><b>Cs</b><br>0.79 | 56<br><b>Ba</b><br>0.89 | 57<br><b>La</b><br>1.10 | 72<br><b>Hf</b><br>1.30  | 73<br><b>Ta</b><br>1.50  | 74<br><b>W</b><br>1.70   | 75<br><b>Re</b><br>1.90  | 76<br><b>Os</b><br>2.20  | 77<br><b>Ir</b><br>2.20  | 78<br><b>Pt</b><br>2.28  | 79<br><b>Au</b><br>2.54  | 80<br><b>Hg</b><br>2.00  | 81<br><b>Tl</b><br>1.62 | 82<br><b>Pb</b><br>2.33 | 83<br><b>Bi</b><br>2.02 | 84<br><b>Po</b><br>2.00 | 85<br><b>At</b><br>2.20 | 86<br><b>Rn</b><br>2.20 |
| 7<br><b>Fr</b><br>0.70 | 88<br><b>Ra</b><br>0.90 | 89<br><b>Ac</b><br>1.10 | 104<br><b>Rf</b><br>1.10 | 105<br><b>Db</b><br>1.10 | 106<br><b>Sg</b><br>1.10 | 107<br><b>Bh</b><br>1.10 | 108<br><b>Hs</b><br>1.10 | 109<br><b>Mt</b><br>1.10 | 110<br><b>Ds</b><br>1.10 | 111<br><b>Rg</b><br>1.10 | 112<br><b>Cn</b><br>1.10 |                         |                         |                         |                         |                         |                         |

**Electron Affinity (kJ/mol)**

| <i>IA</i>               | <i>IIA</i>               | <i>IIIB</i>     | <i>IVB</i>       | <i>VB</i>        | <i>VIIB</i>      | <i>VIIB</i>      | <i>VIIIB</i>     | <i>VIIIB</i>     | <i>IB</i>        | <i>IIB</i>       | <i>IIIA</i>      | <i>IVA</i>      | <i>VA</i>       | <i>VIA</i>      | <i>VIIA</i>     | <i>VIIIA</i>    |                         |
|-------------------------|--------------------------|-----------------|------------------|------------------|------------------|------------------|------------------|------------------|------------------|------------------|------------------|-----------------|-----------------|-----------------|-----------------|-----------------|-------------------------|
| 1                       | 2                        | 3               | 4                | 5                | 6                | 7                | 8                | 9                | 10               | 11               | 12               | 13              | 14              | 15              | 16              | 17              | 18                      |
| 1<br><b>H</b><br>72.77  |                          |                 |                  |                  |                  |                  |                  |                  |                  |                  |                  |                 |                 |                 |                 |                 | 2<br><b>He</b>          |
| 2<br><b>Li</b><br>59.63 | 4<br><b>Be</b><br>0.00   |                 |                  |                  |                  |                  |                  |                  |                  |                  |                  |                 |                 |                 |                 |                 | 5<br><b>B</b><br>26.99  |
| 3<br><b>Na</b><br>52.87 | 12<br><b>Mg</b><br>0.00  |                 |                  |                  |                  |                  |                  |                  |                  |                  |                  |                 |                 |                 |                 |                 | 6<br><b>C</b><br>121.78 |
| 4<br><b>K</b><br>48.38  | 20<br><b>Ca</b><br>2.37  | 21<br><b>Sc</b> | 22<br><b>Ti</b>  | 23<br><b>V</b>   | 24<br><b>Cr</b>  | 25<br><b>Mn</b>  | 26<br><b>Fe</b>  | 27<br><b>Co</b>  | 28<br><b>Ni</b>  | 29<br><b>Cu</b>  | 30<br><b>Zn</b>  | 31<br><b>Ga</b> | 32<br><b>Ge</b> | 33<br><b>As</b> | 34<br><b>Se</b> | 35<br><b>Br</b> | 36<br><b>Kr</b><br>0.00 |
| 5<br><b>Rb</b><br>46.88 | 38<br><b>Sr</b><br>4.63  | 39<br><b>Y</b>  | 40<br><b>Zr</b>  | 41<br><b>Nb</b>  | 42<br><b>Mo</b>  | 43<br><b>Tc</b>  | 44<br><b>Ru</b>  | 45<br><b>Rh</b>  | 46<br><b>Pd</b>  | 47<br><b>Ag</b>  | 48<br><b>Cd</b>  | 49<br><b>In</b> | 50<br><b>Sn</b> | 51<br><b>Sb</b> | 52<br><b>Te</b> | 53<br><b>I</b>  | 54<br><b>Xe</b><br>0.00 |
| 6<br><b>Cs</b><br>45.50 | 56<br><b>Ba</b><br>13.95 | 57<br><b>La</b> | 72<br><b>Hf</b>  | 73<br><b>Ta</b>  | 74<br><b>W</b>   | 75<br><b>Re</b>  | 76<br><b>Os</b>  | 77<br><b>Ir</b>  | 78<br><b>Pt</b>  | 79<br><b>Au</b>  | 80<br><b>Hg</b>  | 81<br><b>Tl</b> | 82<br><b>Pb</b> | 83<br><b>Bi</b> | 84<br><b>Po</b> | 85<br><b>At</b> | 86<br><b>Rn</b><br>0.00 |
| 7<br><b>Fr</b>          | 88<br><b>Ra</b>          | 89<br><b>Ac</b> | 104<br><b>Rf</b> | 105<br><b>Db</b> | 106<br><b>Sg</b> | 107<br><b>Bh</b> | 108<br><b>Hs</b> | 109<br><b>Mt</b> | 110<br><b>Ds</b> | 111<br><b>Rg</b> | 112<br><b>Cn</b> |                 |                 |                 |                 |                 |                         |

|   |                 |                 |                 |                 |                 |                 |                 |                 |                 |                 |                  |                  |                  |                  |
|---|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|------------------|------------------|------------------|------------------|
| 6 | 58<br><b>Ce</b> | 59<br><b>Pr</b> | 60<br><b>Nd</b> | 61<br><b>Pm</b> | 62<br><b>Sm</b> | 63<br><b>Eu</b> | 64<br><b>Gd</b> | 65<br><b>Tb</b> | 66<br><b>Dy</b> | 67<br><b>Ho</b> | 68<br><b>Er</b>  | 69<br><b>Tm</b>  | 70<br><b>Yb</b>  | 71<br><b>Lu</b>  |
| 7 | 90<br><b>Th</b> | 91<br><b>Pa</b> | 92<br><b>U</b>  | 93<br><b>Np</b> | 94<br><b>Pu</b> | 95<br><b>Am</b> | 96<br><b>Cm</b> | 97<br><b>Bk</b> | 98<br><b>Cf</b> | 99<br><b>Es</b> | 100<br><b>Fm</b> | 101<br><b>Md</b> | 102<br><b>No</b> | 103<br><b>Lr</b> |

**First Ionization Energy (kJ/mol)**

| <i>IA</i>               | <i>IIA</i>              | <i>IIIB</i>              | <i>IVB</i>               | <i>VB</i>                | <i>VIB</i>               | <i>VIIB</i>              | <i>VIIIB</i>             | <i>VIIIB</i>             | <i>IB</i>                | <i>IIB</i>                | <i>IIIA</i>              | <i>IVA</i>               | <i>VA</i>                | <i>VIA</i>               | <i>VIIA</i>               | <i>VIIIA</i>              |                           |
|-------------------------|-------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|---------------------------|--------------------------|--------------------------|--------------------------|--------------------------|---------------------------|---------------------------|---------------------------|
| 1                       | 2                       | 3                        | 4                        | 5                        | 6                        | 7                        | 8                        | 9                        | 10                       | 11                        | 12                       | 13                       | 14                       | 15                       | 16                        | 17                        | 18                        |
| 1<br><b>H</b><br>1312.0 |                         |                          |                          |                          |                          |                          |                          |                          |                          |                           |                          |                          |                          |                          |                           |                           | 2<br><b>He</b><br>2372.3  |
| 2<br><b>Li</b><br>520.2 | 3<br><b>Be</b><br>899.5 |                          |                          |                          |                          |                          |                          |                          |                          |                           |                          |                          |                          |                          |                           |                           | 5<br><b>B</b><br>800.6    |
| 3<br><b>Na</b><br>495.8 | 4<br><b>Mg</b><br>737.7 |                          |                          |                          |                          |                          |                          |                          |                          |                           |                          |                          |                          |                          |                           |                           | 6<br><b>C</b><br>1086.5   |
| 4<br><b>K</b><br>418.8  | 5<br><b>Ca</b><br>589.8 | 21<br><b>Sc</b><br>633.1 | 22<br><b>Ti</b><br>658.8 | 23<br><b>V</b><br>650.9  | 24<br><b>Cr</b><br>652.9 | 25<br><b>Mn</b><br>717.3 | 26<br><b>Fe</b><br>762.5 | 27<br><b>Co</b><br>760.4 | 28<br><b>Ni</b><br>737.1 | 29<br><b>Cu</b><br>745.5  | 30<br><b>Zn</b><br>906.4 | 31<br><b>Ga</b><br>578.8 | 32<br><b>Ge</b><br>762.2 | 33<br><b>As</b><br>944.5 | 34<br><b>Se</b><br>941.0  | 35<br><b>Br</b><br>1139.9 | 36<br><b>Kr</b><br>1350.8 |
| 5<br><b>Rb</b><br>403.0 | 6<br><b>Sr</b><br>549.5 | 37<br><b>Y</b><br>599.9  | 38<br><b>Zr</b><br>640.1 | 39<br><b>Nb</b><br>652.1 | 40<br><b>Mo</b><br>684.3 | 41<br><b>Tc</b><br>710.2 | 42<br><b>Ru</b><br>719.7 | 43<br><b>Rh</b><br>804.4 | 44<br><b>Pd</b><br>731.0 | 45<br><b>Ag</b><br>867.8  | 46<br><b>Cd</b><br>558.3 | 47<br><b>In</b><br>708.6 | 48<br><b>Sn</b><br>830.6 | 49<br><b>Sb</b><br>869.3 | 50<br><b>Te</b><br>1008.4 | 51<br><b>I</b><br>1170.3  |                           |
| 6<br><b>Cs</b><br>375.7 | 7<br><b>Ba</b><br>502.9 | 55<br><b>La</b><br>658.5 | 56<br><b>Hf</b><br>658.4 | 57<br><b>Ta</b><br>758.8 | 58<br><b>W</b><br>755.8  | 59<br><b>Re</b><br>814.2 | 60<br><b>Os</b><br>865.2 | 61<br><b>Ir</b><br>864.4 | 62<br><b>Pt</b><br>890.1 | 63<br><b>Au</b><br>1007.1 | 64<br><b>Hg</b><br>589.4 | 65<br><b>Tl</b><br>715.6 | 66<br><b>Pb</b><br>703.0 | 67<br><b>Bi</b><br>869.3 | 68<br><b>Po</b><br>1008.4 | 69<br><b>At</b><br>1170.3 |                           |
| 7<br><b>Fr</b>          | 8<br><b>Ra</b>          | 87<br><b>Ac</b>          | 88<br><b>Rf</b>          | 89<br><b>Db</b>          | 104<br><b>Sg</b>         | 105<br><b>Bh</b>         | 106<br><b>Hs</b>         | 107<br><b>Mt</b>         | 108<br><b>Ds</b>         | 109<br><b>Rg</b>          | 110<br><b>Cn</b>         |                          |                          |                          |                           |                           |                           |

|                |                 |                 |                 |                 |                 |                 |                 |                 |                 |                 |                  |                  |                  |
|----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|------------------|------------------|------------------|
| 6<br><b>Ce</b> | 58<br><b>Pr</b> | 59<br><b>Nd</b> | 60<br><b>Pm</b> | 61<br><b>Sm</b> | 62<br><b>Eu</b> | 63<br><b>Gd</b> | 64<br><b>Tb</b> | 65<br><b>Dy</b> | 66<br><b>Ho</b> | 67<br><b>Er</b> | 68<br><b>Tm</b>  | 69<br><b>Yb</b>  | 70<br><b>Lu</b>  |
| 7<br><b>Th</b> | 90<br><b>Pa</b> | 91<br><b>U</b>  | 92<br><b>Np</b> | 93<br><b>Pu</b> | 94<br><b>Am</b> | 95<br><b>Cm</b> | 96<br><b>Bk</b> | 97<br><b>Cf</b> | 98<br><b>Es</b> | 99<br><b>Fm</b> | 100<br><b>Md</b> | 101<br><b>No</b> | 102<br><b>Lr</b> |

**Empirical Atomic Radii (pm)**

| <i>IA</i>       | <i>IIA</i>       | <i>IIIB</i>     | <i>IVB</i>       | <i>VB</i>       | <i>VIB</i>       | <i>VIIB</i>      | <i>VIIIB</i>     | <i>VIIIB</i>    | <i>IB</i>        | <i>IIB</i>       | <i>IIIA</i>      | <i>IVA</i>       | <i>VA</i>        | <i>VIA</i>       | <i>VIIA</i>      | <i>VIIIA</i>    |                  |
|-----------------|------------------|-----------------|------------------|-----------------|------------------|------------------|------------------|-----------------|------------------|------------------|------------------|------------------|------------------|------------------|------------------|-----------------|------------------|
| 1               | 2                | 3               | 4                | 5               | 6                | 7                | 8                | 9               | 10               | 11               | 12               | 13               | 14               | 15               | 16               | 17              | 18               |
| 1<br>H<br>35    |                  |                 |                  |                 |                  |                  |                  |                 |                  |                  |                  |                  |                  |                  |                  | 2<br>He<br>31   |                  |
| 2<br>Li<br>145  | 4<br>Be<br>105   |                 |                  |                 |                  |                  |                  |                 |                  |                  |                  |                  |                  |                  |                  |                 |                  |
| 3<br>Na<br>180. | 12<br>Mg<br>150. |                 |                  |                 |                  |                  |                  |                 |                  |                  |                  |                  |                  |                  |                  |                 |                  |
| 4<br>K<br>220.  | 20<br>Ca<br>180. | 21<br>Sc<br>164 | 22<br>Ti<br>140. | 23<br>V<br>135  | 24<br>Cr<br>140. | 25<br>Mn<br>140. | 26<br>Fe<br>140. | 27<br>Co<br>135 | 28<br>Ni<br>135  | 29<br>Cu<br>135  | 30<br>Zn<br>135  | 31<br>Ga<br>130. | 32<br>Ge<br>125  | 33<br>As<br>115  | 34<br>Se<br>115  | 35<br>Br<br>115 | 36<br>Kr<br>88   |
| 5<br>Rb<br>235  | 38<br>Sr<br>200. | 39<br>Y<br>180. | 40<br>Zr<br>155  | 41<br>Nb<br>145 | 42<br>Mo<br>145  | 43<br>Tc<br>135  | 44<br>Ru<br>130. | 45<br>Rh<br>135 | 46<br>Pd<br>140. | 47<br>Ag<br>160. | 48<br>Cd<br>155  | 49<br>In<br>155  | 50<br>Sn<br>145  | 51<br>Sb<br>145  | 52<br>Te<br>140. | 53<br>I<br>140. | 54<br>Xe<br>108  |
| 6<br>Cs<br>260. | 56<br>Ba<br>215  | 57<br>La<br>188 | 72<br>Hf<br>155  | 73<br>Ta<br>145 | 74<br>W<br>135   | 75<br>Re<br>135  | 76<br>Os<br>130. | 77<br>Ir<br>135 | 78<br>Pt<br>135  | 79<br>Au<br>135  | 80<br>Hg<br>150. | 81<br>Tl<br>190. | 82<br>Pb<br>180. | 83<br>Bi<br>160. | 84<br>Po<br>190. | 85<br>At<br>127 | 86<br>Rn<br>120. |
| 7<br>Fr<br>215  | 88<br>Ra<br>195  | 89<br>Ac<br>104 | 105<br>Rf<br>105 | 106<br>Db<br>Sg | 107<br>Bh<br>Bh  | 108<br>Hs<br>Mt  | 109<br>Mt<br>Ds  | 110<br>Ds<br>Rg | 111<br>Rg<br>Cn  | 112<br>Cn        |                  |                  |                  |                  |                  |                 |                  |

|   |                  |                  |                 |                 |                  |                 |                  |                 |                 |                 |                  |                  |                  |                  |
|---|------------------|------------------|-----------------|-----------------|------------------|-----------------|------------------|-----------------|-----------------|-----------------|------------------|------------------|------------------|------------------|
| 6 | 58<br>Ce<br>183  | 59<br>Pr<br>183  | 60<br>Nd<br>182 | 61<br>Pm<br>181 | 62<br>Sm<br>180. | 63<br>Eu<br>204 | 64<br>Gd<br>180. | 65<br>Tb<br>178 | 66<br>Dy<br>177 | 67<br>Ho<br>177 | 68<br>Er<br>176  | 69<br>Tm<br>175  | 70<br>Yb<br>194  | 71<br>Lu<br>173  |
| 7 | 90<br>Th<br>180. | 91<br>Pa<br>180. | 92<br>U<br>175  | 93<br>Np<br>175 | 94<br>Pu<br>175  | 95<br>Am<br>175 | 96<br>Cm<br>175  | 97<br>Bk<br>175 | 98<br>Cf<br>175 | 99<br>Es<br>175 | 100<br>Fm<br>175 | 101<br>Md<br>175 | 102<br>No<br>175 | 103<br>Lr<br>175 |

| Covalent Radii (pm)         |                         |                        |                         |                        |                        |                        |                        |                        |                        |                        |                        |                        |                        |                        |                        |                         |                         |
|-----------------------------|-------------------------|------------------------|-------------------------|------------------------|------------------------|------------------------|------------------------|------------------------|------------------------|------------------------|------------------------|------------------------|------------------------|------------------------|------------------------|-------------------------|-------------------------|
| IA                          | IIA                     | IIIB                   | IVB                     | VB                     | VIIB                   | VIIIB                  | VIIIB                  | VIIIB                  | IB                     | IIB                    | IIIA                   | IVA                    | VA                     | VIA                    | VIIA                   | VIIIA                   |                         |
| 1                           | 2                       | 3                      | 4                       | 5                      | 6                      | 7                      | 8                      | 9                      | 10                     | 11                     | 12                     | 13                     | 14                     | 15                     | 16                     | 17                      | 18                      |
| 1<br><b>H</b><br>38         |                         |                        |                         |                        |                        |                        |                        |                        |                        |                        |                        |                        |                        |                        |                        |                         | 2<br><b>He</b><br>28    |
| 2<br>3<br><b>Li</b><br>134  | 4<br><b>Be</b><br>90.   |                        |                         |                        |                        |                        |                        |                        |                        |                        |                        |                        |                        |                        |                        |                         | 5<br><b>B</b><br>82     |
| 2<br>11<br><b>Na</b><br>154 | 12<br><b>Mg</b><br>130. |                        |                         |                        |                        |                        |                        |                        |                        |                        |                        |                        |                        |                        |                        |                         | 6<br><b>C</b><br>77     |
| 3<br>19<br><b>K</b><br>196  | 20<br><b>Ca</b><br>174  | 21<br><b>Sc</b><br>144 | 22<br><b>Ti</b><br>136  | 23<br><b>V</b><br>125  | 24<br><b>Cr</b><br>127 | 25<br><b>Mn</b><br>139 | 26<br><b>Fe</b><br>125 | 27<br><b>Co</b><br>126 | 28<br><b>Ni</b><br>121 | 29<br><b>Cu</b><br>138 | 30<br><b>Zn</b><br>131 | 31<br><b>Ga</b><br>126 | 32<br><b>Ge</b><br>122 | 33<br><b>As</b><br>119 | 34<br><b>Se</b><br>116 | 35<br><b>Br</b><br>114  | 36<br><b>Kr</b><br>116  |
| 4<br>37<br><b>Rb</b><br>211 | 38<br><b>Sr</b><br>192  | 39<br><b>Y</b><br>162  | 40<br><b>Zr</b><br>148  | 41<br><b>Nb</b><br>137 | 42<br><b>Mo</b><br>145 | 43<br><b>Tc</b><br>156 | 44<br><b>Ru</b><br>126 | 45<br><b>Rh</b><br>135 | 46<br><b>Pd</b><br>131 | 47<br><b>Ag</b><br>153 | 48<br><b>Cd</b><br>148 | 49<br><b>In</b><br>144 | 50<br><b>Sn</b><br>141 | 51<br><b>Sb</b><br>138 | 52<br><b>Te</b><br>135 | 53<br><b>I</b><br>133   | 54<br><b>Xe</b><br>140. |
| 5<br>55<br><b>Cs</b><br>225 | 56<br><b>Ba</b><br>198  | 57<br><b>La</b><br>169 | 72<br><b>Hf</b><br>150. | 73<br><b>Ta</b><br>138 | 74<br><b>W</b><br>146  | 75<br><b>Re</b><br>159 | 76<br><b>Os</b><br>128 | 77<br><b>Ir</b><br>137 | 78<br><b>Pt</b><br>128 | 79<br><b>Au</b><br>144 | 80<br><b>Hg</b><br>149 | 81<br><b>Tl</b><br>148 | 82<br><b>Pb</b><br>147 | 83<br><b>Bi</b><br>146 | 84<br><b>Po</b><br>135 | 85<br><b>At</b><br>150. | 86<br><b>Rn</b><br>145  |
| 6<br>87<br><b>Fr</b>        | 88<br><b>Ra</b>         | 89<br><b>Ac</b>        | 104<br><b>Rf</b>        | 105<br><b>Db</b>       | 106<br><b>Sg</b>       | 107<br><b>Bh</b>       | 108<br><b>Hs</b>       | 109<br><b>Mt</b>       | 110<br><b>Ds</b>       | 111<br><b>Rg</b>       | 112<br><b>Cn</b>       |                        |                        |                        |                        |                         |                         |

|                      |                 |                 |                 |                 |                 |                 |                 |                 |                 |                  |                  |                  |                  |  |  |  |  |
|----------------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|------------------|------------------|------------------|------------------|--|--|--|--|
| 6<br>58<br><b>Ce</b> | 59<br><b>Pr</b> | 60<br><b>Nd</b> | 61<br><b>Pm</b> | 62<br><b>Sm</b> | 63<br><b>Eu</b> | 64<br><b>Gd</b> | 65<br><b>Tb</b> | 66<br><b>Dy</b> | 67<br><b>Ho</b> | 68<br><b>Er</b>  | 69<br><b>Tm</b>  | 70<br><b>Yb</b>  | 71<br><b>Lu</b>  |  |  |  |  |
| 7<br>90<br><b>Th</b> | 91<br><b>Pa</b> | 92<br><b>U</b>  | 93<br><b>Np</b> | 94<br><b>Pu</b> | 95<br><b>Am</b> | 96<br><b>Cm</b> | 97<br><b>Bk</b> | 98<br><b>Cf</b> | 99<br><b>Es</b> | 100<br><b>Fm</b> | 101<br><b>Md</b> | 102<br><b>No</b> | 103<br><b>Lr</b> |  |  |  |  |

**van der Waals Radii (pm)**

| <i>IA</i>             | <i>IIA</i>             | <i>IIIB</i>            | <i>IVB</i>       | <i>VB</i>        | <i>VIB</i>       | <i>VIIB</i>      | <i>VIIIB</i>     | <i>VIIIB</i>     | <i>IB</i>        | <i>IIB</i>       | <i>IIIa</i>      | <i>IVA</i>      | <i>VA</i>       | <i>VIA</i>      | <i>VIIA</i>     | <i>VIIIA</i>    |                        |
|-----------------------|------------------------|------------------------|------------------|------------------|------------------|------------------|------------------|------------------|------------------|------------------|------------------|-----------------|-----------------|-----------------|-----------------|-----------------|------------------------|
| 1                     | 2                      | 3                      | 4                | 5                | 6                | 7                | 8                | 9                | 10               | 11               | 12               | 13              | 14              | 15              | 16              | 17              | 18                     |
| 1<br><b>H</b><br>109  |                        |                        |                  |                  |                  |                  |                  |                  |                  |                  |                  |                 |                 |                 |                 |                 | 2<br><b>He</b><br>140. |
| 2<br><b>Li</b><br>182 | 4<br><b>Be</b><br>153  |                        |                  |                  |                  |                  |                  |                  |                  |                  |                  |                 |                 |                 |                 |                 | 5<br><b>B</b><br>192   |
| 3<br><b>Na</b><br>227 | 12<br><b>Mg</b><br>173 |                        |                  |                  |                  |                  |                  |                  |                  |                  |                  |                 |                 |                 |                 |                 | 6<br><b>C</b><br>170.  |
| 4<br><b>K</b><br>275  | 20<br><b>Ca</b><br>231 | 21<br><b>Sc</b><br>211 | 22<br><b>Ti</b>  | 23<br><b>V</b>   | 24<br><b>Cr</b>  | 25<br><b>Mn</b>  | 26<br><b>Fe</b>  | 27<br><b>Co</b>  | 28<br><b>Ni</b>  | 29<br><b>Cu</b>  | 30<br><b>Zn</b>  | 31<br><b>Ga</b> | 32<br><b>Ge</b> | 33<br><b>As</b> | 34<br><b>Se</b> | 35<br><b>Br</b> | 36<br><b>Kr</b>        |
| 5<br><b>Rb</b><br>303 | 38<br><b>Sr</b><br>249 | 39<br><b>Y</b>         | 40<br><b>Zr</b>  | 41<br><b>Nb</b>  | 42<br><b>Mo</b>  | 43<br><b>Tc</b>  | 44<br><b>Ru</b>  | 45<br><b>Rh</b>  | 46<br><b>Pd</b>  | 47<br><b>Ag</b>  | 48<br><b>Cd</b>  | 49<br><b>In</b> | 50<br><b>Sn</b> | 51<br><b>Sb</b> | 52<br><b>Te</b> | 53<br><b>I</b>  | 54<br><b>Xe</b>        |
| 6<br><b>Cs</b><br>343 | 56<br><b>Ba</b><br>268 | 57<br><b>La</b>        | 72<br><b>Hf</b>  | 73<br><b>Ta</b>  | 74<br><b>W</b>   | 75<br><b>Re</b>  | 76<br><b>Os</b>  | 77<br><b>Ir</b>  | 78<br><b>Pt</b>  | 79<br><b>Au</b>  | 80<br><b>Hg</b>  | 81<br><b>Tl</b> | 82<br><b>Pb</b> | 83<br><b>Bi</b> | 84<br><b>Po</b> | 85<br><b>At</b> | 86<br><b>Rn</b>        |
| 7<br><b>Fr</b>        | 88<br><b>Ra</b>        | 89<br><b>Ac</b>        | 104<br><b>Rf</b> | 105<br><b>Db</b> | 106<br><b>Sg</b> | 107<br><b>Bh</b> | 108<br><b>Hs</b> | 109<br><b>Mt</b> | 110<br><b>Ds</b> | 111<br><b>Rg</b> | 112<br><b>Cn</b> |                 |                 |                 |                 |                 |                        |

|   |                 |                 |                 |                 |                 |                 |                 |                 |                 |                 |                  |                  |                  |                  |
|---|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|------------------|------------------|------------------|------------------|
| 6 | 58<br><b>Ce</b> | 59<br><b>Pr</b> | 60<br><b>Nd</b> | 61<br><b>Pm</b> | 62<br><b>Sm</b> | 63<br><b>Eu</b> | 64<br><b>Gd</b> | 65<br><b>Tb</b> | 66<br><b>Dy</b> | 67<br><b>Ho</b> | 68<br><b>Er</b>  | 69<br><b>Tm</b>  | 70<br><b>Yb</b>  | 71<br><b>Lu</b>  |
| 7 | 90<br><b>Th</b> | 91<br><b>Pa</b> | 92<br><b>U</b>  | 93<br><b>Np</b> | 94<br><b>Pu</b> | 95<br><b>Am</b> | 96<br><b>Cm</b> | 97<br><b>Bk</b> | 98<br><b>Cf</b> | 99<br><b>Es</b> | 100<br><b>Fm</b> | 101<br><b>Md</b> | 102<br><b>No</b> | 103<br><b>Lr</b> |