

Balancing Chemical Equations

Worksheet

Choose the correct balanced equation from the given options.

Which of the following is the correctly balanced equation for the reaction between hydrogen and oxygen to form water?

- a) $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$
- b) $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
- c) $\text{H}_2 + 2\text{O}_2 \rightarrow 2\text{H}_2\text{O}$
- d) $2\text{H}_2 + 2\text{O}_2 \rightarrow 2\text{H}_2\text{O}$

Identify the correctly balanced equation for the combustion of methane (CH_4).

- a) $\text{CH}_4 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
- b) $\text{CH}_4 + 2\text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$
- c) $2\text{CH}_4 + \text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$
- d) $\text{CH}_4 + 3\text{O}_2 \rightarrow 2\text{CO}_2 + 2\text{H}_2\text{O}$

Which of the following equations is correctly balanced for the decomposition of calcium carbonate?

- a) $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$
- b) $2\text{CaCO}_3 \rightarrow \text{CaO} + 2\text{CO}_2$
- c) $\text{CaCO}_3 \rightarrow \text{Ca} + \text{C} + \text{O}_2$
- d) $\text{CaCO}_3 \rightarrow \text{CaO} + \text{O}_2$