## Balancing Chemical Equations Worksheet

1. Write the missing reactant and balance the equation for the combustion of ethylene (C<sub>2</sub>H<sub>4</sub>) which produces carbon dioxide and water.

• \_\_\_ 
$$C_2H_4$$
 + \_\_\_  $O_2 \rightarrow$  \_\_\_  $CO_2$  + \_\_\_  $H_2O$ 

2. Iron reacts with sulfuric acid to produce iron(III) sulfate and hydrogen gas.

• -----

3. Given the products of a reaction are potassium chloride and oxygen, identify the likely reactant if it is a decomposition reaction and balance the equation.

• \_\_\_ KCIO<sub>3</sub> 
$$\rightarrow$$
 \_\_l. KCI + \_\_\_ O<sub>2</sub>