

# Balancing Chemical Equations Worksheet

**BALANCE THE FOLLOWING EQUATION:**  $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$

- A. 1, 1, 1  
B. 2, 1, 2  
C. 2, 2, 2  
D. 2, 1, 1

**Which coefficients correctly balance the equation for the combustion of propane:**  $\text{C}_3\text{H}_8 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$ ?

- A. 1, 5, 3, 4  
B. 1, 3, 3, 4  
C. 1, 5, 3, 5  
D. 1, 5, 3, 3

**Balance the equation:**  $\text{NaCl} + \text{AgNO}_3 \rightarrow \text{AgCl} + \text{NaNO}_3$

- A. 1, 1, 1, 1  
B. 1, 2, 2, 1  
C. 2, 2, 2, 2  
D. 2, 1, 1, 2

**What type of reaction is represented by the equation**  $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ ?

- A. Decomposition  
B. Double replacement  
C. Synthesis  
D. Single replacement

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