

## COUNTING ATOMS WORKSHEET

## Part 1: Basic Counting

Identify and count the atoms in these simple compounds.

1) H<sub>2</sub>O (Water)

2) CO<sub>2</sub> (Carbon Dioxide)

## **Part 2: Intermediate Counting**

These compounds include polyatomic ions. Consider how subscripts outside parentheses multiply the atoms inside.

- 1) CaCl<sub>2</sub> (Calcium Chloride)
- 2) Al(OH)₃ (Aluminum Hydroxide)

If you have 4 molecules of iron(III) oxide (Fe₂O₃), how many iron (Fe) and oxygen (O) atoms are there?