Counting Atoms Worksheet

| Name: | Date: | | |
|--|---|----------|---|
| Calculate the total number of atoms in: | | | |
| 1) 5 molecules of H₂O | 2) 3 molecules of C ₆ H ₁₂ O ₆ | | |
| Apply atom counting to more complex formulas. Count the atoms in each of the following | | | |
| 1) C ₆ H ₁₂ O ₆ (Glucose) | 2) CaCO₃ (Calcium Carbona | te) | |
| True or False | | | |
| Question | | ✓ | X |
| A molecule of water (H ₂ O) contains two hydrogen aton | ns and one oxygen atom. | | |
| Carbon dioxide (CO ₂) consists of two carbon atoms and two oxygen atoms | | | |
| Ammonia has one nitrogen atom and three hydrogen atoms. | | | |
| Methane includes one carbon atom and four hydrogen atoms. | | | |
| In sodium chloride, there is one sodium atom for every | one chlorine atom. | | |
| Glucose is made up of 6 carbon atoms, 12 hydrogen a | toms, and 6 oxygen atoms. | | |
| Calcium carbonate includes three oxygen atoms. | | | |
| A molecule of sulfuric acid contains 2 hydrogen atoms oxygen atoms. | s, 1 sulfur atom, and 4 | | |
| Carbon dioxide plays a critical role in the process of p | hotosynthesis. | | |
| Ammonia is primarily used as a building material in co | onstruction projects. | | |