Atomic Structure Worksheet

What is the atomic number of an element with 11 protons?	An atom that has gained or lost electrons is called a?
○ 22○ 11	 Molecule Isotope
 12 23 	 Ion Proton
How do you calculate the mass number of an atom?	Which particle determines the chemical properties of an element?
Protons + Neutrons	O Proton
O Protons + Electrons	O Electron
O Neutrons - Protons	O Neutron
Electrons + Neutrons	
What does the atomic number of an element represent?	Electrons are found in which part of an atom?
O The number of neutrons	
O The total number of protons & neutrons	Energy levels (shells)
O The number of protons	Inside neutrons
The number of atoms in a molecule	 Inside protons